

Chapter (10)



$$(a) \text{ Density} = \frac{\text{mass}}{\text{volume}} \left(\frac{\text{kg}}{\text{m}^3} \right) \quad \frac{\text{g}}{\text{cm}^3}$$

$$\rho_{\text{water}} = 1000 \text{ kg/m}^3$$

$$\text{Specific gravity (x)} = \frac{\rho_x}{\rho_{\text{water}}}$$

$$S.G(x) = 7.5 \Rightarrow \frac{\rho_x}{1000} \Rightarrow 7500 \text{ kg/m}^3$$

$$\rho_x = 3 \rho_y$$

$$\rho = \frac{m}{V}$$

$$S.G_x = \frac{\rho_x}{\rho_w}$$

$$V = 3 \text{ m}^3$$

$$S.G(x) = 7.5 \rightarrow \rho_x = 7500$$

$$\frac{7500}{3} = \frac{m}{3}$$

$$(y) \text{ mass} = ??$$

$$\frac{7500}{3} = \frac{m}{3}$$

$$m = \frac{7500}{3} \text{ kg}$$

$$P = \frac{F}{A} \sin \theta \quad (\text{N/m}^2) \quad (\text{Pascal})$$



60kg
 $2 \text{ feet} = 500 \text{ cm}^2$

$1 \text{ m}^2 = 100 \text{ cm}^2$
 $1 \text{ m}^2 = 10000 \text{ cm}^2$
 $500 \text{ cm}^2 = ?$

$P = \frac{F}{A} = \frac{600}{0.05} = 12000 \text{ Pas}$
 12 kPas

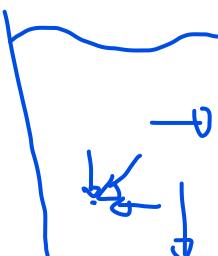
N/m^2
 $F = ??$
 P_1

1 Foot $\rightarrow F ??$
 P_2

$$P = \frac{F}{A} = \frac{600}{0.05} = 24000 \text{ Pas}$$

Static fluids

(1) At any point inside the liquid, the pressure is the same in all directions.

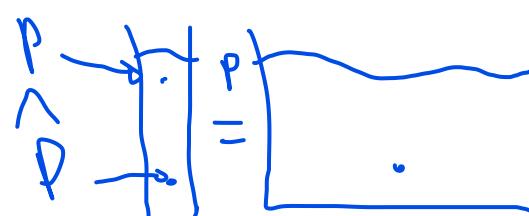
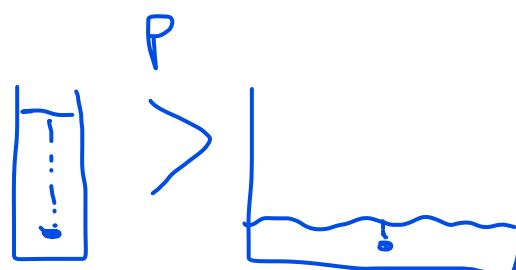


(2) The pressure of any static fluid is always perpendicular to the surface

$$P = \frac{m}{V} \quad (m = \rho V)$$

$$P = \frac{F}{A} = \frac{mg}{A} = \frac{\rho V g}{A} = \frac{\rho A h g}{A}$$

$$P = \rho h g$$

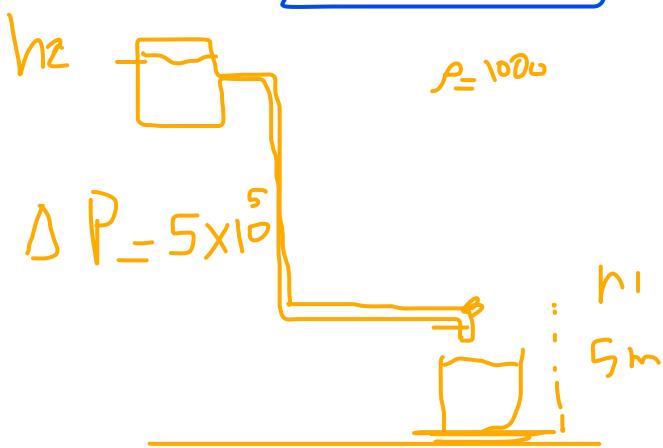
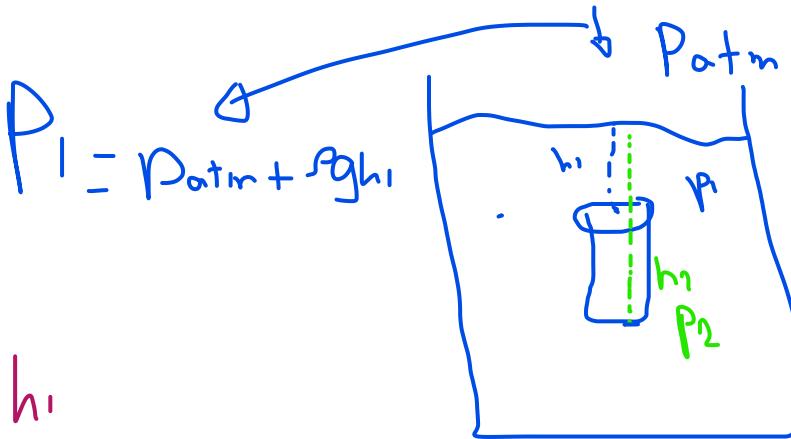


$$P_1 = \rho g h_1$$

$$P_2 = \rho g h_2$$

$$P_2 - P_1 = \underline{\rho g h_2} - \underline{\rho g h_1}$$

$$\Delta P = \underline{\rho g (\Delta h)}$$



$$5 \times 10^5 = 10^3 (10) \Delta h$$

$$5 + 5 \times 10 = h_2$$

$$55 \text{ m} = h_2$$

Atmospheric pressure

$$1.013 \times 10^5 \text{ Pa} = 1.013 \text{ bar} = 1 \text{ atm} = 760 \text{ mm Hg}$$

$$1 \text{ bar} = 10^5 \text{ Pa}$$

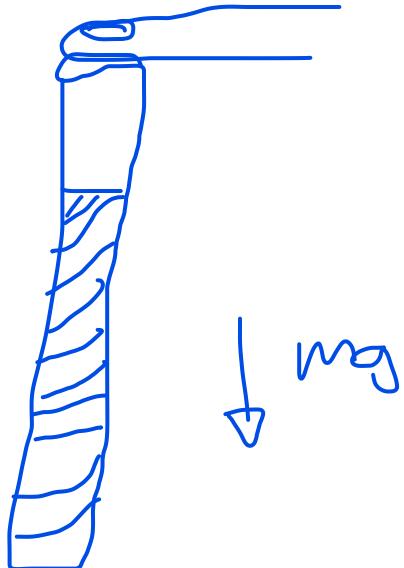
$$760 \text{ mm Hg} = 1 \text{ atm}$$

$$F_{\text{atm}} - mg - F_a = 0$$

$\downarrow F_a$

$$P_{\text{atm}}(A) - \rho V g - PA = 0$$

$$P_{\text{atm}} - \rho g h = 0$$



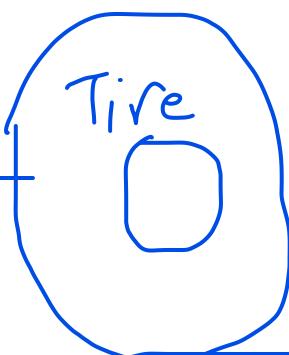
$$P_{\text{atm}} = P + \rho gh$$

$\uparrow F_{\text{atm}}$

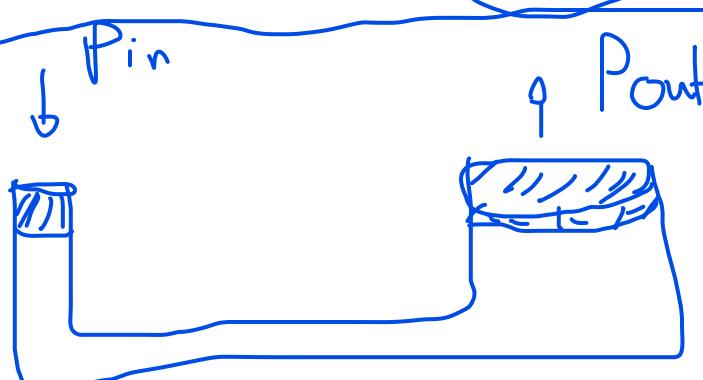
Gauge pressure

Difference between inside and outside (Gauge pressure)

$$P_{\text{atm}} + G.P = \text{pressure in}$$



P_{out}



Hydraulic lift

$$\frac{F_{\text{in}}}{A_{\text{in}}} = \frac{F_{\text{out}}}{A_{\text{out}}}$$

$$\frac{50}{1} = \frac{500}{100}$$

$$P_{atm} + \rho_f g \Delta h = P_{gas}$$

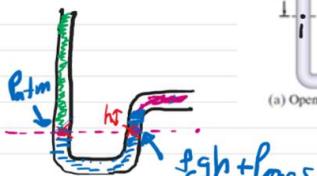
Open-tube manometer

The two points ① and ② are at the same height.

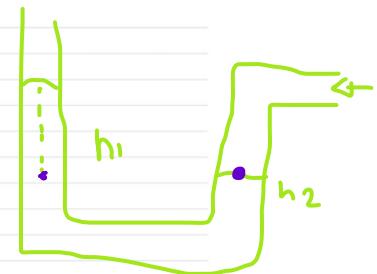
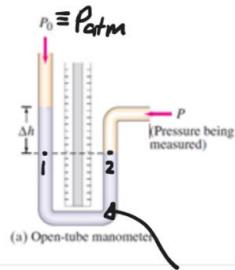
$$P_1 = P_2$$

$$P_{atm} + \rho_f g \Delta h = P_{gas}$$

$$\cdot P_{gas} = P_{atm} + \rho_f g \Delta h$$



$$P_{gas} + \rho_f g h = P_{atm} \Rightarrow P_{gas} = P_{atm} - \rho_f g h$$



$$P_{atm} = P_{gas} + \rho_f g h$$

$$P_{gas} = P_{atm} - \rho_f g h$$

760 mm Hg

Mercury barometer

A column of mercury of height 760 mm (76 cm) results in a pressure equivalent to that of atmospheric pressure.

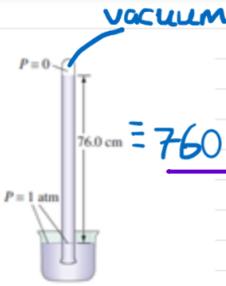
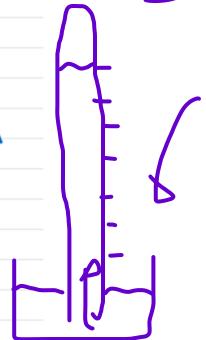


FIGURE 10-8 A mercury barometer, invented by Torricelli, is shown here when the air pressure is standard atmospheric, 76.0 cm-Hg.



Question: If water is used instead of mercury, find the height of the water column to balance the atmospheric pressure.

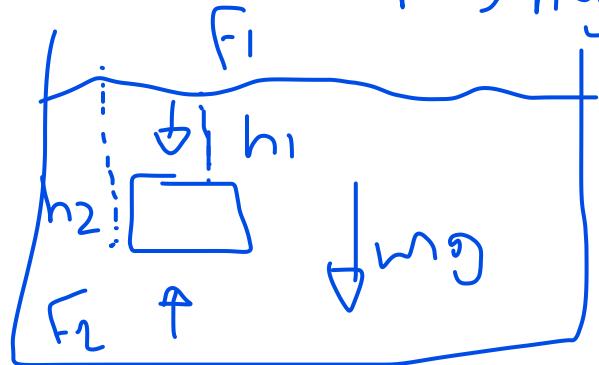
$$\rho V = m$$

Bouyant force

$$F = PA$$

$$P = \rho h g$$

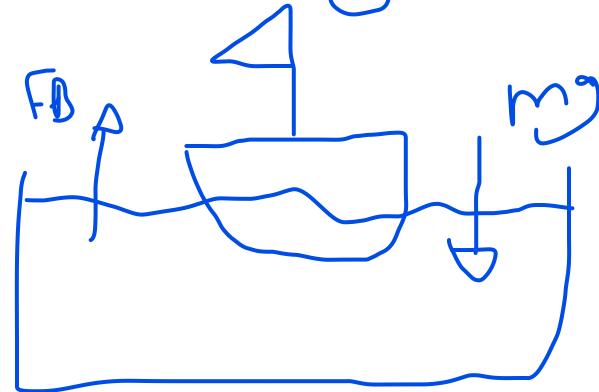
$$F_2 - F_1 = F_B$$



$$F_B = \rho g A h_2 - \rho A g h_1$$

$$F_B = \rho g A h = \rho V g = m g$$

$$\rho_w V g > \rho_s V g$$



$$f_x = 80\% f_y$$



